

EXERCISE - 5

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Q1 Find an exponential growth model $y = y_0 e^{kt}$ that satisfies the stated conditions.

(i) $y_0 = 1$ and doubling time $t = 5$

(ii) $y(0) = 5$ and growth rate = 2%

(iii) $y(1) = 1$ and $y(10) = 100$

(iv) Given growth model is

$$y = y_0 e^{kt}$$

When $t = 0$, $y_0 = 1$

$$y = 1 \times e^0 = 1$$

When $t = 5$, $y = 2y_0$

$$\therefore 2y_0 = y_0 e^{k5}$$

$$2 = e^{5k}$$

$$5k = \log_e 2$$

$$5k = 0.3010 \times 2.303$$

$$k = \frac{0.3010 \times 2.303}{5}$$

$$k = 0.1386$$

$$y = 1 \times e^{0.1386t}$$

$$y = e^{0.1386t}$$

(ii) Equation of growth model is

$$y = y_0 e^{kt} \quad \text{--- (1)}$$

$$t = 0, \quad y = 5$$

$$\therefore 5 = y_0 e^0$$

$$5 = y_0$$

Diff wrt t (1)

$$\frac{dy}{dt} = y_0 k e^{kt}$$

$$\text{at } t=0, \frac{dy}{dt} = \frac{2y_0}{100}$$

$$\frac{2}{100} y_0 = y_0 k e^0$$

$$k = \frac{2}{100}$$

$$y = 5 e^{\frac{2}{100} t} = 5 e^{0.02t}$$

iii)

Equation of growth model is

$$y = y_0 e^{kt}$$

$$\text{when } t=1, y=1$$

$$1 = y_0 e^k \quad \text{--- (1)}$$

$$\text{when } t=10, y=100$$

$$100 = y_0 e^{10k} \quad \text{--- (2)}$$

Divide eqⁿ (2) by (1) $100 = \frac{e^{10k}}{e^k}$

$$100 = e^{9k} \Rightarrow \log_e 100 = 9k$$

$$2 \log_e 10 = 9k$$

$$2 \times 2.303 = 9k \Rightarrow \frac{2 \times 2.303}{9} = k$$

$$0.511 = k \quad \text{Put in (1)}$$

$$1 = y_0 e^{0.511}$$

$$y_0 = \frac{1}{e^{0.511}} = e^{-0.511}$$

$$y = e^{-0.511} e^{0.511t}$$

$$\approx 0.60 e^{0.511t}$$

$$e^{-0.511} = \frac{1}{e^{0.511}} = \frac{1}{(2.718)^{0.511}} \approx \frac{1}{\sqrt{2.718}}$$

$$\approx \frac{1}{1.648}$$

$$\approx 0.60$$

Q3

In a certain culture of bacteria the number of bacteria increased 5 times in 10 hours. How long did it take for the number of bacteria to double.

Solution

Let P be the number of bacteria after time t years

$$\frac{dP}{dt} \propto P$$

$$\frac{dP}{dt} = KP$$

$$\int \frac{dP}{P} = \int K dt$$

$$\log_e P = Kt + C$$

$$P = e^{Kt+C}$$

$$P = e^{Kt} e^C$$

$$P = A e^{Kt} \quad (\text{where } A = e^C) \quad \text{--- (1)}$$

Let at $t=0$, $P = P_1$

Put in (1)

$$P_1 = A e^0$$

$$A = P_1$$

$$P = P_1 e^{Kt}$$

ATQ at $t=10$, $P = 5P_1$

$$\therefore 5P_1 = P_1 e^{10K}$$

$$5 = e^{10K}$$

$$10K = \log_e 5$$

$$10K = 1.61$$

$$K = 0.161$$

$$P = P_1 e^{0.161t}$$

Let at $t=T$, $P=2P_1$

$$\therefore 2P_1 = P_1 e^{0.161T}$$

$$\log_e 2 = 0.161T$$

$$0.6931 = 0.161T$$

$$T = \frac{0.6931}{0.161}$$

$$T = 4.3 \text{ hr}$$

\therefore No. of bacteria will be double in 4.3 hours.

Q11 Use the exponential growth model to show that the time it takes for a population to double (i.e. from an initial number A to $2A$) is given by $t = \frac{\ln 2}{k}$.

Solution Let P be the population at time t .

$$\therefore \frac{dP}{dt} \propto P$$

$$\frac{dP}{dt} = kP$$

$$\int \frac{dP}{P} = \int k dt$$

$$\log P = kt + C$$

$$P = e^{kt+C}$$

$$P = A e^{kt} \quad (\text{where } e^C = A)$$

Let at $t = 0$, $P = A$ (initially)

$$\therefore A = A e^{k \cdot 0}$$

$$A = A$$

$$\therefore P = A e^{kt}$$

Let at time t , $P = 2A$

$$\therefore 2A = A e^{kt}$$

$$2 = e^{kt}$$

$$\log_e 2 = kt$$

$$t = \frac{\log_e 2}{k}$$

Q4

The amount of oil pumped from one of the wells decreases at the continuous rate of 10% per year. When will the well's output fall to one-fourth of its present value?

Solution

Let (P) be the oil pumped in time (t)
Then A.T.S

$$\frac{dP}{dt} = \frac{-10P}{100}$$

$$\int \frac{dP}{P} = \frac{-1}{10} \int dt$$

$$\log_e P = \frac{-1}{10} t + C$$

$$P = e^{\frac{-1}{10}t + C}$$

$$= e^{\frac{-1}{10}t} \cdot e^C$$

$$P = \lambda e^{\frac{-1}{10}t} \quad (\text{where } e^C = \lambda)$$

Let initially at $t=0$, $P=P_1$

$$\boxed{P_1 = \lambda}$$

$$\text{Hence } P = P_1 e^{\frac{-1}{10}t}$$

Now let at $t=T$, $P = \frac{1}{4} P_1$

$$\therefore \frac{1}{4} P_1 = P_1 e^{\frac{-1}{10}T}$$

$$\frac{1}{4} = e^{-\frac{1}{10}T}$$

$$\log_e \left(\frac{1}{4} \right) = -\frac{1}{10}T$$

$$-\log_e 4 = -\frac{1}{10}T$$

$$2 \log_e 2 = \frac{1}{10}T$$

$$2 \times 0.6931 = \frac{1}{10}T$$

$$2 \times 10 \times 0.6931 = T$$

$$13.8620 = T$$

Wells output fall to one-fourth in 13.8 years.

Q8 Half-life of radioactive carbon-14 is 5700 years. A ~~the~~ certain bone was observed to contain 75% of carbon-14 as compared to what is present in the living creatures. Determine its antiquity.

Solution Let (P) be the mass of carbon-14 at time (t)

$$\therefore \frac{dP}{dt} \propto (-P) \Rightarrow \frac{dP}{dt} = -kP$$

$$\int \frac{dP}{P} = -k \int dt$$

$$\int \log_e P = -kt + C$$

$$P = e^{-kt+C}$$

$$P = e^{-kt} e^C$$

$$P = A e^{-kt} \quad (\text{where } e^0 = 1)$$

Let at $t = 0$ (initially), $P = P_0$

$$P_0 = A e^0$$

$$A = P_0$$

$$P = P_0 e^{-kt}$$

at $t = 5700$ years, $P = \frac{P_0}{2}$

$$\frac{P_0}{2} = P_0 e^{-5700k}$$

$$\frac{1}{2} = e^{-5700k}$$

$$\log \frac{1}{2} = -5700k$$

$$-\log 2 = -5700k$$

$$0.6931 = 5700k$$

$$k = \frac{0.6931}{5700}$$

$$P = P_0 e^{-\frac{0.6931}{5700}t}$$

Let at $t = T$, $P = \frac{75}{100} P_0$

$$\frac{75}{100} P_0 = P_0 e^{-\frac{0.6931}{5700}T}$$

$$\frac{3}{4} = e^{-\frac{0.6931}{5700}T}$$

$$\log_e \frac{3}{4} = -\frac{0.6931}{5700}T$$

$$-0.2876 = \frac{-0.6931T}{5700}$$

$$\frac{0.2876 \times 5700}{0.6931} = T$$

$$2365.19 = T$$

Antiquity of bone is 2365.19 years.

82 Gaurav deposited ₹ 5000 in an account paying 3% interest compounded continuously for 5 years.

(i) Find the total amount at the end of 5 years

(ii) How long will it take for the money to double?

Solution Let (A) be the amount at time t compounded continuously at rate $r\%$

$$\frac{dA}{dt} = \frac{r}{100} A$$

$$\int \frac{dA}{A} = \frac{r}{100} \int dt$$

$$\log A = \frac{rt}{100} + C$$

$$A = e^{\frac{rt}{100} + C}$$

$$A = e^{\frac{rt}{100}} e^C$$

$$A = A_0 e^{rt/100} \quad \text{--- (6)}$$

$$\text{at } t=0, \quad A = ₹ 5000$$

$$\therefore 5000 = A_0$$

Thus $A = 5000 e^{x/100}$ — (2)

$$A = 5000 e^{0.03t}$$

(i) at $t = 5$
 Amount $A = 5000 e^{0.03 \times 5}$
 $= 5000 e^{0.15}$
 $= 5000 \times 1.1618$
 $= \underline{\underline{5809.17}}$

(ii) Let at $t = T$, money will double
 ∴ from equation (2)

$$2A_0 = A_0 e^{0.03T}$$

$$2 = e^{0.03T}$$

$$\log_e 2 = 0.03T$$

$$\Rightarrow 0.6931 = 0.03T$$

$$T = \frac{0.6931}{0.03} = 23.10 \text{ years}$$

Q5

A cup of tea with temperature 95°C placed in a room with a constant temperature of 21°C . How many minutes will it take to reach a temperature of 51°C if it cools to 85°C in 1 minute.

Solution Let T be the temperature of tea at time (t)

∴ By Newton's Law of Cooling
 The rate of change of temp (T) of body is proportional to the difference between (T) and the

temperature of surrounding medium A

$$\frac{dT}{dt} \propto (T-A)$$

$$\frac{dT}{dt} = -R(T-A)$$

$$\int \frac{dT}{T-A} = -R \int dt$$

$$\log(T-A) = -kt + C$$

$$\log(T-21) = -kt + C$$

$$T-21 = e^{-kt+C}$$

$$T-21 = \lambda e^{-kt}$$

$$T-21 = \lambda e^{-kt} \quad \text{--- (1)}$$

where $\lambda = e^C$

at $t=0$, $T=95$

\therefore from (1)

$$95-21 = \lambda$$

$$\lambda = 74$$

$$T-21 = 74e^{-kt} \quad \text{--- (2)}$$

when $t=1$ min, $T=85^\circ\text{C}$

from (2)

$$85-21 = 74e^{-k}$$

$$\frac{64}{74} = e^{-k}$$

$$e^k = \frac{74}{64}$$

$$e^k = \frac{74}{64}$$

$$k = \log \frac{74}{64}$$

$$R = 0.1452$$

Put in (2)

$$T-21 = 74e^{-0.1452t}$$

when $T=51$

$$51-21 = 74e^{-0.1452t}$$

$$30 = 74 e^{-0.1452t}$$

$$30 = 74 e^{-0.1452t}$$

$$\frac{30}{74} = e^{-0.1452t}$$

$$0.1452t = \log_e \left(\frac{74}{30} \right)$$

$$0.1452t = 0.90286$$

$$t = \frac{0.90286}{0.1452}$$

$$t = 6.2 \text{ min}$$

$$t = 6.2 \text{ min}$$

Q7

Radium decomposes at a rate proportional to the amount present.

If half the original amount disappears in 1600 years, find

the percentage lost in 100 years.

Solution

Let (P) be the amount of Radium at time (t) years

ATQ

$$\frac{dP}{dt} \propto (-P)$$

$$\frac{dP}{dt} = -kP$$

$$\int \frac{dP}{P} = -k \int dt$$

$$\log_e P = -kt + C$$

$$P = e^{-kt+C}$$

$$P = e^{-kt} e^C$$

$$P = \lambda e^{-kt} \quad (\lambda = e^C)$$

at $t = 0$, let $P = P_0$

$$P_0 = \lambda e^0$$

$$A = P_0$$

$$P = P_0 e^{-kt}$$

ATQ at $t = 1600$ years

$$P = \frac{P_0}{2}$$

$$\frac{P_0}{2} = P_0 e^{-1600k}$$

$$\frac{1}{2} = e^{-1600k}$$

$$-1600k = \log 1 - \log 2$$

$$+ 1600k = + 0.6931$$

$$k = \frac{0.6931}{1600}$$

$$P = P_0 e^{\frac{0.6931 \times 1600}{1600}}$$

at $t = 100$

$$P = P_0 e^{-\frac{0.6931 \times 100}{1600}}$$

$$P = P_0 e^{-\frac{0.6931}{16}}$$

$$\frac{P_0}{P} = e^{\frac{0.6931}{16}}$$

$$\log \left(\frac{P_0}{P} \right) = \frac{0.6931}{16}$$

$$\log_e \left(\frac{P_0}{P} \right) = 0.04331$$

$$\frac{P_0}{P} = e^{0.04331}$$

$$\frac{P_0}{P} = 1.0442$$

$$\frac{P}{P_0} = \frac{1}{1.0442}$$

$$\frac{P}{P_0} = 0.9577$$

$$P = 0.9577 P_0$$

Radium lost in 100 years

$$= P_0 - 0.9577 P_0$$

$$= 0.0423 P_0$$

$$= \frac{0.0423 P_0}{P_0} \times 100\%$$

$$= 4.2\%$$